

TOTAL SYNTHESIS OF (+)-ELDANOLIDE FROM D-RIBONOLACTONE

R. M. Ortuño, R. Mercé and J. Font*

Departament de Química (Divisió de Química Orgànica), Facultat de Ciències
Universitat Autònoma de Barcelona, Bellaterra (Barcelona), Spain.

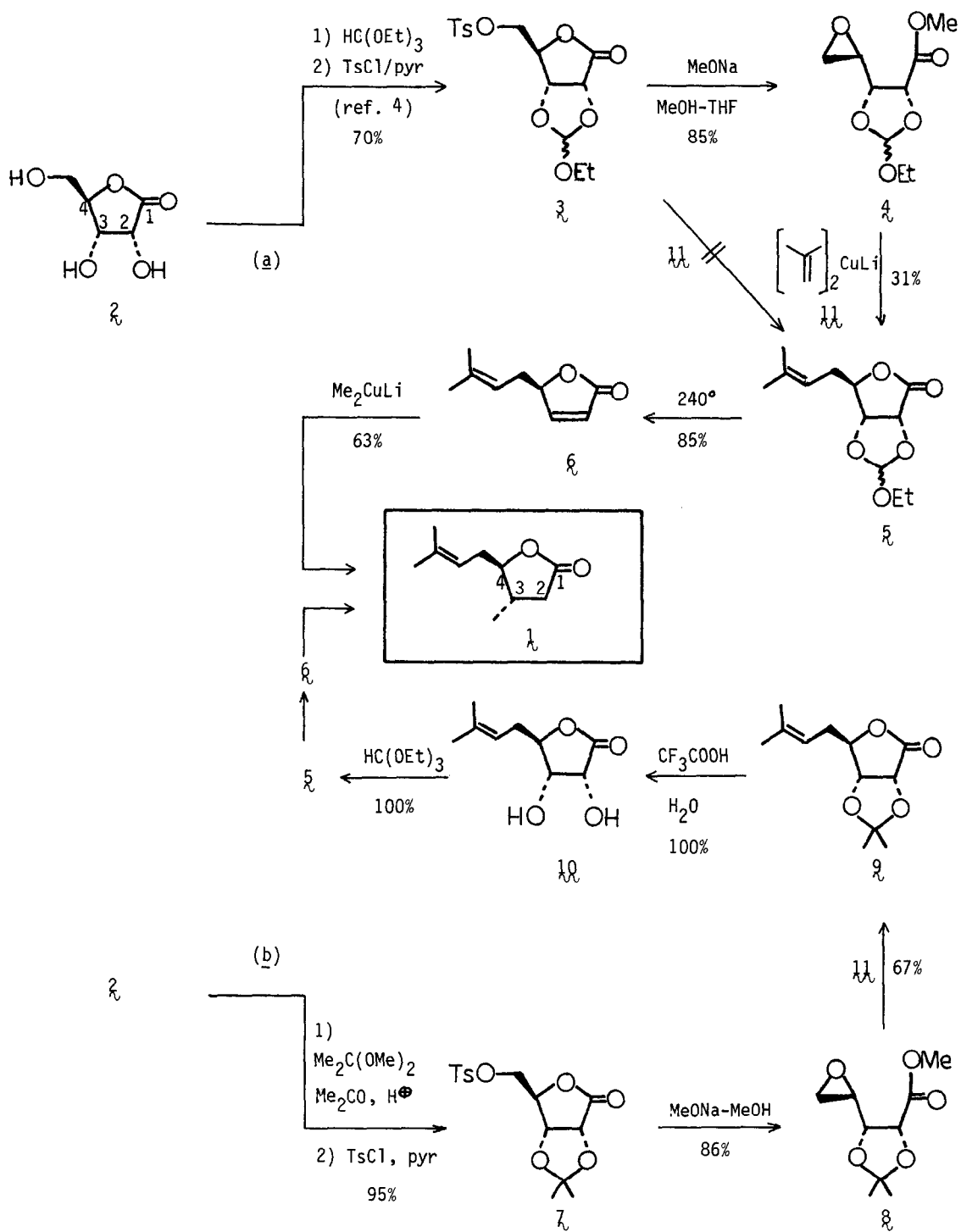
Abstract.- A new and useful synthesis of (+)-Eldanolide is reported, D-ribonolactone being used as a chiral starting product.

(3S,4R)-(+)-Eldanolide, λ , is a sex attractant pheromone, isolated from the male wing glands of the African sugar cane borer Eldana Saccharina (Wlk)¹. The structure and absolute configuration assignment and a total synthesis of this natural product have been reported by Vigneron and Larchevêque et al². Other syntheses of Eldanolide in enantiomerically pure and racemic forms have also been published by other authors, as a consequence of a general interest on this target molecule³.

We describe herein a new and convenient synthetic pathway to (+)-Eldanolide, λ , using D-ribonolactone, ζ , as a chiral starting product.

D-Ribonolactone, ζ , has been proved to be a useful precursor in the preparation of many interesting chiral compounds, such as ranunculin⁴, some pheromones⁵, litsenolides⁶ and (-)-blastmycinone⁷. In our synthetic strategy chirality at C-4 of ζ is transferred to C-4 of the target λ , through the sequences (a and b) depicted in the Scheme. In these convergent routes, the butenolide δ is a common key precursor, that leads to λ by a stereospecific Michael addition of Me_2CuLi . Indeed, conjugate additions of nucleophiles to γ -substituted α,β -butenolides are reported to occur in a highly stereocontrolled manner, giving trans-3,4-disubstituted butyrolactones⁸. In this sense, chirality at C-4 of ζ is also responsible for the asymmetry induced at C-3 when a Michael addition is performed on butenolide δ .

Preparation of λ according to sequence a involves conversion of D-ribonolactone, ζ , into the known tosylate orthoester ζ^4 , that yielded by treatment with MeONa at 0° the new epoxide ζ , b.p. 89°/0.5 torr⁹. Compound ζ is the intermediate to obtain the butenolide δ , since opening of the oxirane ring by nucleophilic attack of a cuprate and subsequent lactonization could lead to



Scheme

the isopentenyl side-chain formation, while pyrolysis of the orthoester would allow the creation of the conjugate C-C double bond of $\mathfrak{6}$. Thus, reaction of epoxide $\mathfrak{4}$ with lithium diisobutenyl cuprate¹⁰, $\mathfrak{11}$, afforded $\mathfrak{5}$, b.p. 120°/1 torr, that, by heating at 240°, gave (R)-(-)-7-methyl-2,6-octadiene-4-olide², $\mathfrak{6}$, $\{\alpha\}_D^{20} = -132.8^\circ$ (c=1.56, MeOH), in 85 % yield. Finally, addition of Me_2CuLi led to (+)-Eldanolide, $\mathfrak{1}$, as a liquid, b.p. 120°/12 torr, $\{\alpha\}_D^{20} = +50.7^\circ$ (c=2.8, MeOH), in 63 % yield. (Lit.² $\{\alpha\}_D^{20} = +51.5^\circ$ (c=1.15, MeOH)).

Yields of the transformations performed in this synthetic sequence are quite good, except in step $\mathfrak{4} \rightarrow \mathfrak{5}$. We have already verified in our laboratory that reactions of tosylate $\mathfrak{3}$ with lithium dialkylcuprates give only poor yields, probably due to steric hindrance effects, and this is the reason we carried out the reaction on epoxide $\mathfrak{4}$ instead of on tosylate $\mathfrak{3}$. Moreover, we have reported in an earlier publication⁵ that the orthoester function is not compatible with the reaction conditions involved in this kind of transformation, and leads to the formation of undesired by-products and polymeric material resulting in low yields.

With these points in mind, we designed the alternative sequence b. D-Ribonolactone, $\mathfrak{2}$, was converted into the tosylate acetonide $\mathfrak{7}^{11}$, that led to epoxide $\mathfrak{8}$, $\{\alpha\}_D^{20} = -14.5^\circ$ (c=2.4, CH_2Cl_2) in 86 % yield. Reaction of $\mathfrak{8}$ with lithium diisobutenyl cuprate, $\mathfrak{11}$, afforded $\mathfrak{9}$, $\{\alpha\}_D^{20} = -34.1^\circ$ (c=1.7, CH_2Cl_2), in 67 % yield; $\mathfrak{9}$ was quantitatively hydrolyzed¹² to the diol $\mathfrak{10}$, m.p. 103-104°, $\{\alpha\}_D^{20} = +33.5^\circ$ (c=1.0), MeOH), and orthoesterification of $\mathfrak{10}$ led to $\mathfrak{5}$, converging with the former route. The longer synthetic route b is more convenient because of the improved yield in the epoxide opening step ($\mathfrak{8}$ vs $\mathfrak{4}$) with cuprate $\mathfrak{11}$. On the other hand, both additional transformations -acetonide formation and hydrolysis- were achieved almost quantitatively. The overall yield in (+)-Eldanolide, $\mathfrak{1}$, obtained through sequence b is 30 % from D-ribonolactone, being the best yield reported till now, starting from a commercially available compound.

Studies on syntheses of other interesting products, using D-ribonolactone derivatives as chiral precursors, are in progress¹³.

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Notes and References

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